GEM-ACCEPTOR, DONOR-DISUBSTITUTED DIENES IN CYCLOADDITION FACILE [1,3] REARRANGEMENT OF NEW BICYCLO [2.2.1] HEPT-5-ENE-7-ONES

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Summary: The (4+2) cycloaddition of methyl 1-diethylamino-2-ethyl-1,3-butadiene-1-carboxylate was performed with acrylonitrile. The cycloadduct bridges over to form 1-diethylamino-6-ethyl-bicyclo [2.2.1] hept-5-ene-7-one-2-carbonitrile 3. We demonstrate that this latter product undergoes facile [1,3] signatropic rearrangement.

There is a continued interest in being able to synthesize new aminocarboxylates for pharmacological study. In a previous note¹, we reported the original synthesis of aminobicyclo [2.2.1] heptenones 3 and aminobicyclo [3.2.0] heptenones 4. These unexpected compounds are formed when dienaminocarboxylates 1 are mixed with acrylonitrile under thermal cycloaddition conditions. The cycloadduct 2 is not isolated.



We now report a mechanistic study of the action of methyl 1-diethylamino-2-ethyl-1,3-butadiene-1carboxylate 1 with acrylonitrile. X-ray, ¹H and ¹³C NMR analyses give the regio-and stereoisomery of the new bicycloheptenonecarbonitriles 3 and 4. This structural assignment offers a new insight into the mechanism of the formation of bicycloheptenones. The dienaminocarboxylate 1 (0.05 mol.) was reacted with acrylonitrile (large excess, reflux 36h). The 1-diethylamino-6-ethyl bicyclo [2.2.1] hept-5-en-7-one-2-carbonitrile 3 and the 3-diethylamino-4ethyl bicyclo [3.2.0] hept-3-en-2-onecarbonitrile 4 were isolated (50/50) by preparative HPLC.

A single crystal X-ray analysis³ is performed on aminobicyclo [2.2.1] heptenone 3. A drawing of the molecular structure is shown in figure 1, and shows an endo-carbonitrile stereoisomery.



Previously², we have identified compounds 3 and 4.

Figure 2



The regio- and stereochemical assignment of the aminobicyclo [3.2.0] heptenone 4 is made from 1 H and 13 C NMR.

Although difficulties have often been encountered⁴ in attempts to assign the stereochemistry of the four-membered ring system, unambiguous stereochemical assignment can however be made in our aminobicyclo [3.2.0] heptenonecarbonitrile from the vicinal coupling constants between protons of the cyclobutane ring (figure 2). Indeed, we know that the two junction cycle protons must have a cisrelationship, ${}^{3}J_{cis}H_{1}H_{4} = 6.5$ Hz. It can be seen that one of the junction protons (H₄) shows a cisrelationship with proton (H_{3x}) and a trans-relationship with proton (H_{3n}). It is known⁴ that a very small value of ${}^{3}J$ in the four-membered ring is indicative of a trans-relationship between two vicinal protons. The relationship ${}^{3}J_{cis} > {}^{3}J_{trans}$ is observed in all types of four-membered ring. Accordingly, the values of the vicinal coupling constants reported in figure 2 are ${}^{3}J_{cis} = 11.5$ Hz and ${}^{3}J_{trans} = 4.0$ Hz.

Figure 3.

The position of the cyano group in the aminobicyclo [3.2.0] heptenone 4 is found by analyzing its ¹³C NMR spectrum (90 MHz, CHCl₃). The C=O signal at 206.5 ppm comprising a quintuplet (figure 3) involves four equivalent couplings (J \approx 4.0 Hz each). It is known⁵ that ²J O=<u>C</u>-C-<u>H</u> assumes values between 5 and 8 Hz, and ³J_{syn} and ³J_{anti} O=<u>C</u>-C-C-<u>H</u> have similar values. The signal multiplicity is not an artefact, since a ¹³C NMR spectrum (67.88 MHz, CHCl₃) shows the same multiplicity. Only the regioisomery shown in scheme 1 permits these couplings.

Thus, the following transformations may be considered :

Scheme 1.



We initially thought that aminobicyclo [2.2.1] heptenone 3 and aminobicyclo [3.2.0] heptenone 4 would result from [4+2] and [2+2] cycloaddition reactions of aminocyclopentadienone 5. However, we now show that aminocyclopentadienone is not a reactional intermediary in the transformations studied. Indeed, aminocyclopentadienone 5 is not formed by cyclisation of dienaminocarboxylate 1 (scheme 1e).

IR, NMR spectra and gas chromatography did not reveal the formation of methanol when heating 1 at 90°C for 72 hours. The aminocyclopentadienone 5 would not lead to (scheme 1d) this site selectivity by $\pi^2 + \pi^2$ cycloaddition⁶. Moreover, after heating (96h) a solution of aminobicyclo [2.2.1] heptenone 3 in a large excess of acrylonitrile at 90°C, aminobicyclo [3.2.0] heptenone 4 is quantitatively isolated. It results that 4 is very certainly formed by [1,3] signatropic rearrangement of aminobicyclo [2.2.1] heptenone 3 (scheme 1c) with retention of configuration in migrating group. This is according to a typical pericyclic process⁷ in which the migration framework and the migrating group have widely different polarities, specially when the migrating group is substituted by a cyano group.

This study points to the following conclusions :

a- Cycloaddition (4+2) between dienaminocarboxylate 1 and acrylonitrile has taken place.

b- The aminocyclohexenecarboxylate 2 is not isolated; his cyclisation to aminobicyclo [2.2.1] heptenone 3 may be explained by the basic environment and steric hindrance.

c- The [1,3] sigmatropic rearrangement of the aminobicyclo [2.2.1] heptenone 3 leads to conjugate aminobicyclo [3.2.0] heptenone 4.

REFERENCES AND NOTES

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- 3. The crystals of methyl 1-diethylamino-6-ethyl bicyclo [2.2.1] hept-5-ene-7-one-2-carboxylate are monoclinical, space group P2₁/c, a = 7.774(5), b = 15.321(7), c = 11.556(4) Å. $\rho_c = 1.14$ g.cm⁻³ for 4 formula units per cell. A total of 2295 independent reflexions were collected of which 1144 were accepted as statistically above background on the basis that I $\geq 3\sigma(I)$. The structure was solved by direct methods using the Multan-80 program (P.Main, S.J.Fiske, S.E.Hull, L.Lessinger, G.Germain, J.P.Declercq, M.M.Woolfson, Multan 80, Univ.York England and Louvain Belgium, 1980) and refined to a final R = 0.0694.
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